

Project 1022519

## **Integrated Investigation on the Production and Fate of Organo-Cr(III) Complexes from Microbial Reduction of Chromate**

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**RESULTS TO DATE:** Progress and immediate actions for each task are summarized here.

### **Task 1. Production of soluble organo-Cr(III) complexes by selected microorganisms**

The screening of different genera of bacteria for production of soluble Cr(III) complexes has been completed. A total of eight organisms were screened for production of soluble Cr(III); three were Gram positive and five were Gram negative. The Gram positive bacteria were *Cellulomonas* sp. ES 6, *Rhodococcus* sp., and *Leafsonia* sp., while *Shewanella oneidensis* MR 1, *Desulfovibrio desulfuricans* G20, *D. vulgaris* Hildenborough, *Pseudomonas putida* MK 1 and *Ps. aeruginosa* PAO 1 were Gram negative. *S. oneidensis* MR 1 and *Cellulomonas* sp. ES 6 were grown in minimal media, GWM (Ground Water Medium with lactate/fumarate) and SGM ( Simulated Groundwater Medium with sucrose), respectively. Other bacteria were screened under non-growth conditions with sucrose, lactate, or glycerol as electron donor. All experiments were carried out for a period of 15-30 days, with different organisms reaching a maximum soluble Cr(III) concentrations at different times: *S. oneidensis*, 2d; *Cellulomonas* sp., 8d; *Leafsonia*, 6d; *Rhodococcus*, 9d; *Ps. putida* MK 1, 6d, *Ps. aeruginosa* PAO 1, 3d; *D. vulgaris* Hildenborough, 3d; and *D. desulfuricans* G20, 21d. Initial characterization indicates that the soluble Cr(III) fraction produced by both *S. oneidensis* MR 1 and *Cellulomonas* sp. ES 6 passes through a 1-Kd cut off filter. Thus the Cr(III) does not appear to be protein bound under these conditions. The soluble Cr complex of *S. oneidensis* MR 1 was enriched by lyophilization followed by solvent extraction. After this purification about 12.5% of the soluble Cr was soluble in methanol, and about 60% of the complex was soluble in water. The water soluble fraction was retained on an anion exchange column and could be eluted by a salt solution, indicating the Cr(III) is in a complex with negative charges. Further characterization of this compound is underway, and a similar approach will be attempted to characterize the soluble Cr(III) complexes produced by the other bacteria.

We have been developing techniques for characterizing the organo-Cr(III) complexes. We have developed a method of mass spectrometry to enable us to characterize Cr(III) complexes with simple organic ligands. With our method, we are attempting to characterize the complexation of Cr(III) with gluconate, ascorbate, and NADH. Once this method is fully developed, then we will attempt to evaluate the organo-Cr(III) complexes produced during microbial reduction.

We are also developing electron paramagnetic resonance spectrometry (EPR) methods to follow the oxidation state of the Cr during reduction from chromate to Cr(III). Recent work by others has indicated the possibility of identifying intermediates such as Cr(V) during reduction to Cr(III). Although this oxidation state is only present as an intermediate, it may play a role in formation of the final organo-Cr(III) complexes.

**Task 2. Demonstrate that chromate reduction produces organo-Cr(III) complexes with microbial cellular components.** Research focused on the production of soluble organo-Cr(III) complexes has proceeded using a method of reducing 5 mM Cr(VI) in the presence of 20 mM organic in a highly buffered solution at neutral pH, 100 mM Potassium Phosphate (pH 7). We have tested for the formation of soluble organo-Cr(III) complexes using 18 different organic components, amino acids, TCA cycle intermediates, etc. Thus far we have identified 10 highly soluble organo-Cr(III) complexes, 2 slightly soluble complexes with both soluble and insoluble Cr(III), and 6 that form mainly insoluble Cr(III) complexes. Five different highly soluble complexes have been analyzed by UV/vis spectrometry and EPR. Results of the UV/vis analysis showed a red shifting of the Cr(III) peak in the 580 - 600nm range. Similar results were noted for the

previously published NAD<sup>+</sup>-Cr(III) complexes and ascorbate-Cr(III) complexes. EPR analysis gave peaks with much broader curves in comparison to the monomeric Cr(III) control. The broad spectra were again similar to both the NAD<sup>+</sup>-Cr(III) and ascorbate Cr(III) complexes. Thus far the results indicate that soluble organo-Cr(III) complexes readily form in the presence of cellular organics. Further experiments to characterize the complexes based on the organic to Cr(III) ratio of the complex, molecular weight by mass spectrometry, and identifying the specific ligands involved in binding the organic portion to the Cr(III) ion will be performed.

Task 3. investigate the biological and abiotic transformation of organo-Cr(III) complexes formed during microbial reduction of chromate.

Investigation into the biological transformation of NAD<sup>+</sup>-Cr(III) complexes by the gram-positive bacterium has thus far identified that NAD<sup>+</sup> is the carbon source for growth. This has been confirmed by repeated growth experiments using NAD<sup>+</sup>-Cr(III) complexes as the sole carbon and energy source in solution, and by growing the bacterium on a minimal medium using NAD<sup>+</sup> as the sole carbon and energy source. Further characterization of the insoluble Cr(III) after NAD<sup>+</sup> consumption by the bacterium will be performed by transmission electron microscopy coupled with an electron X-ray dispersive spectrophotometer, to gain a better understanding of the nature of the Cr(III) precipitate. Preliminary phylogenetic analysis of the 16 s rDNA identifies the bacterium as belonging to the *Leifsonia* genus, which are gram-positive soil aerobes. Future experiments to characterize the biochemical and metabolic profiles of the bacterium will be completed along with screening the bacterium for mineralization of other organo-Cr(III) complexes.

Task 4. Transport and fate of organic-Cr(III) complexes in soil. Batch equilibrium soil sorption experiments using artificially produced organo Cr(III) complexes (Cr(III)-serine, -cysteine and -malate) will be performed. These experiments will be followed up with soil sorption studies of the purified compounds obtained directly from bacterial cultures. This task was delayed as efforts were focused on Task 1, 2, and 3 for the first year.

**DELIVERABLES:** We have not published any papers, yet. We have two manuscripts in preparation.